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Term	Documents
IN.DWPI,EPAB,JPAB,USPT,PGPB.	179799
INS.DWPI,EPAB,JPAB,USPT,PGPB.	29435
SITU.DWPI,EPAB,JPAB,USPT,PGPB.	130002
SITUS.DWPI,EPAB,JPAB,USPT,PGPB.	3235
(1 SAME (IN ADJ SITU)).USPT,PGPB,JPAB,EPAB,DWPI.	0

Search History

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Database: IBM Technical Disclosure Bulletins

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DB Name	Query	Hit Count	Set Name
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USPT,PGPB,JPAB,EPAB,DWPI 11	same (fix\$ or cross link)	2	<u>L2</u>
USPT,PGPB,JPAB,EPAB,DWPI g	glutaraldehyde bisulfite	110	<u>L1</u>

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Search Results - Record(s) 1 through 2 of 2 returned.

_ 1. Document ID: US 5811529 A

L2: Entry 1 of 2

File: USPT

Sep 22, 1998

US-PAT-NO: 5811529

DOCUMENT-IDENTIFIER: US 5811529 A

TITLE: Bis-pyridone compounds

DATE-ISSUED: September 22, 1998

INVENTOR - INFORMATION:

NAME

CITY

STATE ZIP CODE COUNTRY

Vines; Danette R.

Charlotte

NC

Pedemonte: Ronald P.

Eppstein-Vockenhausen

DEX

US-CL-CURRENT: 534/560; 534/561

ABSTRACT:

A fiber reactive dye having the formula ##STR1## wherein R.sub.1 and R.sub.3 are independently selected from the group consisting of; hydrogen; an alkyl group having from one to four carbon atoms; the foregoing alkyl group further substituted with one or more groups selected from hydroxy; amino; sulfo; halogen; an aryl radical; a heterocyclic radical or combination thereof; where for R.sub.3 said amino group maybe further substituted by a nitrogen heterocyclic fiber reactive group of the series: 1, 3, 5 mono or dichloro triazinyl; 1, 3, 5 mono or difluoro triazinyl; trichloropyrimidinyl; difluoropyrimidinyl; or monochlorodifluoro pyrimidinyl; where the nitrogen heterocycle may be further substituted by an: alkyl; or aryl amino group;

wherein B is selected from the group consisting of a C.sub.2 to C.sub.4 alkyl chain; a substituted aryl or a heterocyclic radical;

wherein X is selected from the group consisting of SO.sub.3 H; SO.sub.3 Na or a hydroxy group; and wherein Q.sub.1 and Q.sub.2 are independently selected from diazo components.

8 Claims, 0 Drawing figures Exemplary Claim Number: 1

L2: Entry 1 of 2

File: USPT

Sep 22, 1998

DOCUMENT-IDENTIFIER: US 5811529 A TITLE: Bis-pyridone compounds

BSPR:

The structure of the coupling component of the present invention incorporates advantages over the prior art. For example, this coupling agent includes a bridge, whereby two solubilizing groups are incorporated in the dyestuff molecule by one condensation reaction. Not only does the bridge enhance solubility through the sulfonic acid moieties present, but it also provides a new way to incorporate two mono-reactive dyes producing one larger dye with improved levels of fixation. In the case of the aliphatic bridge, such as the glutaraldehyde bisulphite condensation product, the fixation properties should be further enhanced. The improved solubility should make these new dyes less prone to precipitate out of solution upon the introduction of salt and alkali. Further, the bridge links two reactive chromophoric systems together which should in theory provide greater tinctorial strength.

Full Title Citation Front Review Classification Date Reference Claims KMC Draw Desc Image

L2: Entry 2 of 2

File: USPT

May 5, 1981

US-PAT-NO: 4266010

DOCUMENT-IDENTIFIER: US 4266010 A

TITLE: Silver halide photographic light-sensitive material

DATE-ISSUED: May 5, 1981

INVENTOR - INFORMATION:

NAME CIT

CITY

STATE ZIP CODE C

COUNTRY

Nagatomo; Shigeru

Minami-ashigara

JPX

Hori; Kiyotaka

Minami-ashigara

JPX

US-CL-CURRENT: 430/355; 430/539, 430/543, 430/642, 430/950, 430/961

ABSTRACT:

A silver halide photographic light-sensitive material containing at least one photographic layer containing acid-processed gelatin and a matting agent.

10 Claims, 0 Drawing figures Exemplary Claim Number: 1

L2: Entry 2 of 2

File: USPT

May 5, 1981

DOCUMENT-IDENTIFIER: US 4266010 A

TITLE: Silver halide photographic light-sensitive material

DETL:

Processing Step Development 25 seconds

Fixation 25 seconds Washing 20 seconds Developer Composition Sodium Sulfite 40 g

Hydroquinone 25 g Boric Acid 10 g 1-Phenyl-3-pyrazolidone 1.5 g Potassium

Hydroxide 30 g 5-Methylbenzotriazole 0.15 g Glutaraldehyde-bisulfite 15 g Acetic

Acid 12 g Potassium Bromide 5 g Water to make 1 l Fixing Solution Ammonium

Thiosulfate 174 g Sodium Sulfite (anhydrous) 20 g Sodium Tetraborate

(decahydrate) 20 g Acetic Acid 25 g Sulfuric Acid 5 g Aluminum Sulfate 7 g Water

to make 1 l

Full Title Citation Front Review Classification Date Reference Claims (MMC Draw Desc Image)

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=> s glutaraldehyde bisulfite

L1 3 GLUTARALDEHYDE BISULFITE

> DUPLICATE PREFERENCE IS 'MEDLINE KEEP DUPLICATES FROM MORE THAN ONE FILE? PROCESSING COMPLETED FOR L1 2 DUPLICATE REMOVE L1 (1 DUPLICATE REMOVED) => d 1-2 bib ab 1.2 ANSWER 1 OF 2 BIOSIS COPYRIGHT 2001 BIOSIS AN 2000:400550 BIOSIS DN PREV200000400550 TIDisinfectant composition. ΑU Synodis, Joseph; Wilensky, Stuart (1); Halecky, Alan CS (1) Matawan, NJ USA ASSIGNEE: Block Drug Company, Inc. рT US 6034138 March 07, 2000 50 Official Gazette of the United States Patent and Trademark Office Patents, (Mar. 7, 2000) Vol. 1232, No. 1, pp. No pagination. e-file. ISSN: 0098-1133. Patent DT LA English AB

The present invention comprises a concentrated solid or semi-solid disinfectant or sterilant composition for use in an aqueous disinfecting or sterilizing solution, comprising an oxidant and a protected glutaraldehyde such as a glutaraldehyde bisulfite addition compound (GBS): ##STR1## or a glutaraldehyde dioxime compound (GDO): ##STR2## The present invention further provides a method for disinfecting or sterilizing a surface or apparatus comprising the steps of mixing a concentrated solid or semi-solid glutaraldehyde sterilant composition comprising an oxidizing compound and a protected sterilant with water to form a solution and bringing the solution into contact with the surface or apparatus.

L2ANSWER 2 OF 2 MEDLINE DUPLICATE 1 AN96190982 MEDLINE DN 96190982 PubMed ID: 8604152 TΤ Inactivation of glutaraldehyde by reaction with sodium bisulfite. Jordan S L; Russo M R; Blessing R L; Theis A B ΑU Union Carbide Corporation, Bound Brook, NJ 08805, USA. CS JOURNAL OF TOXICOLOGY AND ENVIRONMENTAL HEALTH, (1996 Feb 23) 47 (3) SO 299-309. Journal code: KAA; 7513622. ISSN: 0098-4108. CYUnited States DT Journal; Article; (JOURNAL ARTICLE) LA English FS Priority Journals EM199605

ED

Entered STN: 19960524

Last Updated on STN: 19960524

Entered Medline: 19960516 AR The microbiocidal activity of glutaraldehyde was inactivated by reaction with sodium bisulfite via formation of a proposed glutaraldehyde -bisulfite complex. High-performance liquid chromatography (HPLC) analysis of 2% (0.2M) alkaline glutaraldehyde indicated complete loss of glutaraldehyde at a 2.2:1 molar ratio of sodium bisulfite to glutaraldehyde. Neither 1.7% (0.17 M) sodium bisulfite alone nor the glutaraldehyde-bisulfite complex was microbiocidal when tested against Escherichia coli, Pseudomonas aeruginosa, Enterobacter aerogenes, and Polybac Polyseed BOD seed inoculum. Bacterial inhibition tests indicated that the glutaraldehyde-sodium bisulfite complex had no effect on the growth of sewage microorganisms at concentrations as high as 50-100 ppm (5 \bar{x} 10(-4)-1 \bar{x} 10(-3) M), with an IC50 of 230-440 ppm (2.3 \bar{x} 10(-3)-4.4 x 10(-3) M), based on glutaraldehyde concentration. A 28-close bottle test showed a 5-d biodegradation of 48% and 51%, and a 15-d biodegradation of 57% and 63% for 3:1 and 2.2:1 bisulfite to glutaraldehyde molar ratios, respectively. Acute aquatic toxicity testing with Daphnia magna demonstrated an LC50 of 41-109 ppm (4.1 x 10(-4)-10.9 x 10(-4) M) and a no-observed-effect concentration (NOEC) of 16 ppm (1.6 x 10(-4) M) for the proposed glutaraldehyde-bisulfite complex (based on glutaraldehyde concentration), approximately 10-fold

higher than found for glutaraldehyde alone, indicating that the proposed

ex is less toxic to the glutaraldehyde-bisulfite co environment than glutaraldehyde.

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     Bone marrow one step fixation-decalcification in Lowy FMA
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     solution: an immunohistological and in situ hybridization study.
ΑU
     Gaulier A; Fourcade C; Szekeres G; Pulik M
CS
     Service d'Anatomie Cytologie Pathologiques, C. H. Victor Dupouy,
     Argenteuil, France.
SO
     PATHOLOGY, RESEARCH AND PRACTICE, (1994 Dec) 190 (12) 1149-61.
     Journal code: PBZ; 7806109. ISSN: 0344-0338.
CY
     GERMANY: Germany, Federal Republic of
DT
     Journal; Article; (JOURNAL ARTICLE)
LA
    English
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     Priority Journals
EΜ
     199507
     Entered STN: 19950807
     Last Updated on STN: 19960129
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AΒ
     The immunoreactivity of paraffin embedded bone marrow biopsies (BMB) was
     studied following a one step 20-hour-fixation-decalcification in
     Lowy formalin mercuric chlorid acid solution which permits
     excellent histological stainings. Antibodies reactive with myeloid,
     megakaryocytic, erythroid cells, T and B lymphocytes, mastocytes and
     metastatic cells were compared. Nearly all antibodies working on paraffin
     sections were demonstrated on Lowy FMA fixed BMB. Special care
     was taken to define an optimal working dilution. Trypsinization was not
     necessary. A slide microwave pre-treatment appeared essential before
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- testing CD20 L26, CD8, CD3, 34, MB1 Kappa and Lambda antibles. It was suitable for UCHL1, LN2, CD30 antibodies. The same fixative allowed an m RNA Kappa or Lambda in myeloma and EBER 1 EBV RNAs in HIV lymphoma visualization by in situ hybridization. The safety handling of the toxic mercuric chloride component is discussed.
- L4 ANSWER 2 OF 2 BIOSIS COPYRIGHT 2001 BIOSIS
- AN 1995:166744 BIOSIS
- DN PREV199598181044
- TI Bone marrow one step fixation-decalcification in **lowy** FMA solution: An immunohistological and in situ hybridization study.
- AU Gaulier, A. (1); Fourcade, C.; Szekeres, G.; Pulik, M.
- CS (1) Serv. d'Anat. Cytol. Pathol., C.H. Victor Dupouy, 69 rue du Lt-Cl Prudhon, 95107 Argenteuil Cedex France
- SO Pathology Research and Practice, (1994) Vol. 190, No. 12, pp. 1149-1161. ISSN: 0344-0338.
- DT Article
- LA English
- The immunoreactivity of paraffin embedded bone marrow biopsies (BMB) was studied following a one step 20-hour-fixation-decalcification in Lowy formalin mercuric chloride acid solution which permits excellent histological stainings. Antibodies reactive with myeloid, megakaryocytic, erythroid cells, T and B lymphocytes, mastocytes and metastatic cells were compared. Nearly all antibodies working on paraffin sections were demonstrated on Lowy FMA fixed BMB. Special care was taken to define an optimal working dilution. Trypsinization was not necessary. A slide microwave pre-treatment appeared essential before testing CD20 L26, CD8, CD3, CD34, MB1 Kappa and Lambda antibodies. It was suitable for UCHL1, LN2, CD30 antibodies. The same fixative allowed an mRNA Kappa or Lambda in myeloma and EBER 1 EBV RNAs in HIV lymphoma visualization by in situ hybridization. The safety handling of the toxic mercuric chloride component is discussed.

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